

SYNTHESIS, CHARACTERIZATION AND CATALYTIC PERFORMANCE OF BIMETALLIC Ir-Pt CATALYSTS FOR TETRALIN HYDROGENATION

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Bimetallic Pt-Pd catalysts have received considerable attention, because they show high activity in a variety of catalytic applications (1) and stability compared with monometallic Pt or Pd catalysts. For instance, the bimetallic Pt-Pd catalysts have been shown to have higher resistance toward poisons compared to Pt catalysts (2-4). For supported bimetallic clusters, the surface and bulk composition strongly depend on a series of parameters, for example, preparation procedures, metal-metal and metal-support interactions. The final goal is to find the optimal proportion of each metal in order to be more active in these processes. In this case a series of Ir and Pt modified SBA-15 were prepared by the consecutive wet impregnation method, with different content of each metal. The final catalysts were characterized by XRD, TEM and H₂ chemisorption to study the effect of content of metal on the dispersion of iridium/platinum. The catalysts synthesized with similar percentage of both metal showed the best activity measured in tetralin hydrogenation.

Platinum (Pt) and Iridium (Ir) nanoparticles were incorporated into SBA-15 support [1-4] by the wet co-impregnation method. The metal precursors were Iridium acetylacetonate and chloroplatinic acid. The samples were calcined at 500°C and previous to the catalytic test the samples were reduced by heating at 2°C/min to 450°C in a H₂ stream for 5 h. Hereinafter this catalyst will be referred as Ir-Pt- SBA-15 (x), with x= 1-3.

The catalytic activity was measured in a 4563Parr reactor, at 250°C, 15 atm of pressure of hydrogen and 360 rpm for tetralin hydrogenation (feed consisted in 50 mL of 5% v/v of tetralin (98.5% FLUKA) in Dodecane).

The Pt-Ir-SBA-15 catalyst with 1 %wt. of each metals had the highest activity measured in tetralin hydrogenation at mild conditions. The good activity was correlated with higher Ir/Pt dispersion on SBA-15 mesostructured material used as support, as we seen by TEM and H₂ Chemisorption, with higher active metal sites exposed to reactant. The Ir-Pt-SBA-15 activity is sufficiently high to envisage use in the final stages of a refinery process producing diesel fuel of high Cetane Number by hydrodearomatization.

Keywords: Iridium-platinum containing SBA-15; Hydrogenation; TEM

References:

1. V. Ponc, G.C. Bond, *Catalysis by Metals and Alloys*, 95, Elsevier, Amsterdam, (1995) p. 734.
2. J. Lee, H. Rhee, *J. Catal.* 177, (1998)
3. T. Matsui, M. Harada, K.K. Bando, M. Toba, Y. Yoshimura, *J. Jpn. Petrol. Inst.* 47, (2004)
4. B. Pawelec, R. Mariscal, R.M. Navarro, S. Van Bokhorst, S. Rojas, J.L.G. Fierro, *Appl. Catal. A: Gen.* 225, (2002)

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